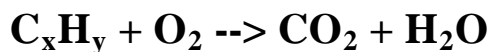




## TYPES OF REACTIONS – HINTS/HELPS

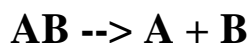
**Combustion** – Oxygen is a reactant and carbon dioxide and water are products.



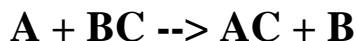
**Synthesis** – TWO OR MORE compounds combine to form ONE compound.



**Decomposition** – ONE larger compound breaks down to make SEVERAL SMALLER ones.



**Single Displacement** – Only ONE element is 'swapped out'.



**Double Displacement** – TWO elements are 'swapped out'.

